

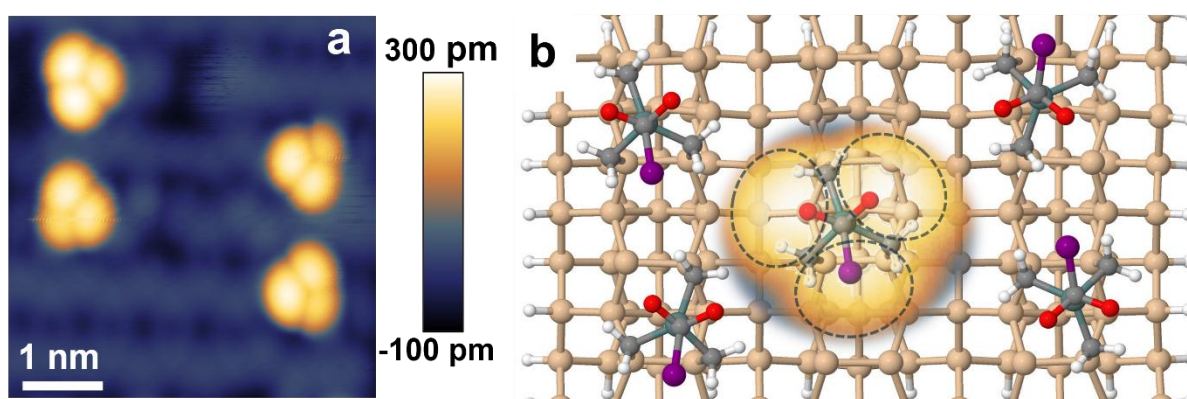
# Ge(CH<sub>2</sub>I)<sub>4</sub> on Si(100): Generating a tip-facing carbon radical for non-planar chemistry

R. Addou, D. G. Allis, J. Barton, B. Blue, A. Bottomley, D. Cheng, N. Culum, M. Drew, T. Enright, A. Godfrey, R. Groome, A. Hill, T. Huff, R. J. Kirby, M. Krykunov, S. Lilak, H. Ma, C. J. Mackie, O. MacLean, T. McCallum, M. Morin, J. Myall, R. Plumadore, A. Powell, H. Rodriguez, L. Sandoval, M. Savoie, B. Scheffel<sup>#</sup>, M. Taucer, D. A. Therien, D. Vobornik, J. Wong

CBN Nano Technologies Inc., 975 Gladstone Ave., K1Y 4W5, Ottawa, Canada

<sup>#</sup> Presenting author's e-mail: bscheffel@nfcbn.com

Scanning probe microscopy (SPM) investigations of on-surface synthesis have garnered significant attention, enabling the formation of synthetically inaccessible off-surface products. However, such studies have largely been confined to high-surface-mobility substrates such as metals or passivated silicon. Here, we expand this paradigm to include reactive silicon surfaces through the development of silicon-specific molecular “tools.” These surface-bound molecules, following a post-deposition activation step, present a tip-facing carbon-centered radical with future applications as either an acceptor of functional molecular units or a donor of small molecular “building” blocks. These tools offer potential for both macroscale and nanoscale applications, including use as a silicon–carbon base with a customizable top layer, or as a carbon fragment donor for tip-mediated mechanosynthesis. To guide the creation of a broader class of such molecular tools, we define six molecular design criteria that enable reproducibility, surface specificity, and experimental verifiability. These are experimentally demonstrated using a highly symmetric model compound: tetrakis(iodomethyl)germane (Ge(CH<sub>2</sub>I)<sub>4</sub>; TIME-Ge). TIME-Ge consists of four iodomethyl (CH<sub>2</sub>I) “legs” attached to a germanium core and was found—under temperature-controlled conditions—to bind to Si(100) 91% of the time with three legs covalently attached and the fourth oriented normal to the surface. Dehalogenation of the tip-facing CH<sub>2</sub>I to form a carbon-centered radical was induced via photochemical and tip-mediated processes. Experimental results were supported by scanning tunneling microscopy (STM), atomic force microscopy (AFM), X-ray photoelectron spectroscopy (XPS), and density functional theory (DFT) calculations. With this framework, a broad and diverse range of new capabilities for molecular engineering are enabled on silicon.



**Figure 1. TIME-Ge molecules on the Si(100) surface at 77 K. (a)** Constant current filled states STM image of four TIME-Ge molecules on Si(100)-(2x1) ( $I = 50$  pA,  $V = -2$  V). **(b)** The four expected on-dimer leg rotational configurations of surface-bound TIME-Ge, predicted by DFT. An experimental tri-lobe is overlaid on a ball-and-stick model in the center, highlighting bright lobes (dashed ellipses) that correlate with the iodine steric minima and the dark lines between lobes correspond to the C-Ge bonds of the underlying 3D scaffold of the molecule.